



117 N. Division St.
Ann Arbor, MI 48104
734-761-3186 x115

To: House Health Policy Committee
Representative Angerer, Chair

From: Michigan Network for Children's Environmental Health
Genevieve Howe (734-663-2400 x115, gen@ecocenter.org)
Brad van Guilder (734-663-2400 x114, bradv@ecocenter.org)

Date: June 21, 2007

Re: Oral Testimony on House Bills 4132 and 4399

Thank you Representative Angerer and the Committee for the opportunity to provide testimony in support of the package of bills before you today.

My name is Gen Howe, and I am here on behalf of the Michigan Network for Children's Environmental Health. The Network is a coalition of health professional, health affected and environmental organizations working to improve the environmental health of Michigan's children. Our Network membership includes the Michigan Chapter of the American Academy of Pediatrics, the Learning Disabilities Association, the Michigan Council for Maternal and Child Health, the Ecology Center, Clean Water Action, the Michigan Environmental Council, and many other groups.

We are pleased the committee is advancing these bills to protect the health of Michigan's children.

As you know, lead remains a critical issue. Michigan currently ranks as the sixth highest state in terms of the estimated population of children with lead poisoning, and the percentage of children found in Michigan with elevated blood lead levels remains higher than the national average, although recent initiatives by the legislature and the Governor have made significant progress.

Lead poisoning is a serious environmental illness that has life-long effects on the individuals who become lead poisoned, and yet is entirely preventable. Lead poisoning in children may affect their health and cognitive abilities, causing permanent and irreversible damage. The lead that accumulates in a child's body and brain may cause anemia, hearing loss, hyperactivity, aggressive behavior, liver and kidney damage, developmental delay, difficulty with learning due to loss of IQ, brain damage, and in extreme cases, even coma and death. Even small levels of lead can be damaging to young children.

Because of this, the Michigan Network for Children's Environmental Health strongly supports House Bills 4132 and 4399. This legislation takes an important first step toward protecting children from lead in children's products.

Unfortunately, children's products can contain toxic heavy metals and other contaminants because our nation's chemical laws do not adequately regulate these chemicals. In 2005, the U.S. General Accounting Office noted that the EPA does not have sufficient authority to regulate potentially hazardous chemicals, and therefore had regulated very few in the last 30 years. Michigan needs to take action where the federal government has failed.

We applaud this first step by the Committee and urge consideration of one amendment to HB 4132 (which would also apply to HB 4399). Given that there is no safe level of lead, we urge the Committee to consider supporting graduated reductions in allowable lead levels in children's products.

We suggest the current legislation include an effort to reduce the allowable lead content of children's products *below* the current standard of 0.06% lead of the total weight (600ppm). The Consumer Product Safety Commission (CPSC) issued an interim enforcement policy for lead in children's jewelry on February 3, 2005 following the tragic death of a young child in Minnesota from acute lead poisoning after swallowing a jewelry pendant. The CPSC policy set an interim enforcement level at 0.06% lead of total weight as a trigger for product recall. The policy also stated "The Commission has urged manufacturers generally to reduce the lead content of their products to the greatest extent possible." The need to reduce lead levels is reinforced by the CPSC's Guidance Policy in Federal Statute 16 C.F.R. Part 1500.230.

An earlier version of the legislation before the Committee included specific requirements for advancing the recommendation of the Consumer Product Safety Commission by reducing allowed levels of lead from 600ppm to 200ppm or 0.02%. We support reinstating this reduction in levels of allowed lead.

In addition, we urge the Committee to consider addressing a broad range of products designed for use or contact by children through companion legislation to that currently before the committee. Product testing has demonstrated that a wide variety of children's products contain lead and other chemicals of concern.

Ideally, legislation to protect children should go beyond lead to include other chemicals that scientific research has demonstrated as known or suspected of interfering with children's developing neurological, hormonal, and reproductive systems. We recommend the best way to do so is to regulate specific groups of chemicals that can pose hazards. (For example, we recommend starting with metals as a group).

We encourage the Committee to regulate known toxicants in children's products using the best practices established elsewhere. We believe the strictest standards possible are important to protect children.

Thank you again for the opportunity to testify, and thank you for your important efforts to protect children's health. Please see our additional detailed comments attached.



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Genevieve Howe (734-663-2400 x115, gen@ecocenter.org)
Brad van Guilder (734-663-2400 x114, bradv@ecocenter.org)

Date: June 21, 2007

Re: Written Testimony on House Bills 4132 and 4399.

Thank you for the opportunity to provide testimony before the House Health Policy Committee regarding House Bills 4132 and 4399. We believe these bills are very important in advancing the protection of children from lead in certain consumer products intended for use by children. We recommend one specific amendment to House Bill 4132 and offer additional comments related to HB 4132 as suggestions for the development of additional legislation.

Overall Comments

We applaud the House Health Policy Committee for proposing legislation on toxicants in children's products.

- We urge the Committee to consider addressing a broad range of products designed for use or contact by children through companion legislation to that currently before the committee. Product testing has demonstrated that a wide variety of children's products contain lead and other chemicals of concern.
- Ideally, legislation to protect children should go beyond lead to include other chemicals that scientific research has demonstrated as known or suspected of interfering with children's developing neurological, hormonal, and reproductive systems. We recommend the best way to do so is to regulate specific groups of chemicals that can pose hazards. (For example, we recommend starting with metals as a group).
- We encourage the Committee to regulate known toxicants in children's products using the best practices established elsewhere. We believe the strictest standards possible are important to protect children.

House Bill 4132

Reducing levels of Lead in Children's products

The current legislation should include an effort to reduce the allowable lead content of children's products *below* the current standard of 0.06% lead of the total weight (600ppm). There is no safe level of lead exposure. The Consumer Product Safety Commission (CPSC) issued an interim enforcement policy for lead in children's jewelry on February 3, 2005 following the tragic death of a young child in Minnesota from acute lead poisoning after swallowing a jewelry pendant. The CPSC policy set an interim enforcement level at 0.06% lead of total weight as a trigger for product recall. The policy also stated "The Commission has urged manufacturers generally to reduce the lead content of their products to the greatest extent possible." The need to reduce lead levels is reinforced by the CPSC's Guidance Policy in Federal Statute 16 C.F.R. Part 1500.230.

An earlier version of the legislation before the committee included specific requirements for advancing the recommendation of the Consumer Product Safety Commission by reducing allowed levels of lead from 600ppm to 200ppm or 0.02%. We support reinstating this reduction in levels of allowed lead.

Reducing levels of lead to 200ppm has also been recommended in several pieces of legislation introduced around the United States. One of the most recent of these is S5784 in the state of New York (attached), which concerns lead in children's jewelry. This bill also includes other features that could help to inform legislation adopted in Michigan.

We suggest page 2, lines 18-22 be modified to read:

(D) Lead-bearing substance means an item or substance that contains lead, or a coating on an item that contains lead, so that the lead content

- (i) On or before August 30, 2009 is more than 0.06% (600 ppm) of the total weight.
- (ii) Beginning August 31, 2009 is more than 0.02% (200 ppm) of the total weight.

Lead-bearing content does not include glass or crystal decorative components.

Definition of Child or Children:

The current legislation defines "Children" as 7 years old or younger. While children 7 years old or younger are of greatest concern we suggest for future legislation that "Children" be defined as individuals who are age 15 years or younger. Toxic metals such as lead are potent neurotoxins that impact brain development. Brain development continues well past age 15. Including these older children in legislation will also help reduce the accessibility of products with toxic content to younger siblings. Federal standards in Canada under "The Children's Jewellery Regulations" effective on May 10, 2005 uses the definition of "a child under the age of 15 years."

Glass or crystal decorative components:

The addition of lead is used to make the components sparkle and it is assumed that the lead is bound in the glass or crystal. However, it seems prudent to ensure that all manufacturers are making a quality product such that items as glass or crystal components are subject to a leachability test. Based on a number of existing standards for leachability of lead we would suggest a mandatory standard of 90 ppm.

ASTM F 963 - 96a: US voluntary standard for toys is 90 ppm

EN 71-3: 1994: European safety standard for toys is 90 ppm

P.C. 2005-805: Canadian Children's Jewellery Regulations standard is 90 ppm.

New York bill S5784 provides for specific testing protocols for glass or crystal decorative components as well as other components and materials common in jewelry intended for children. We urge the committee to provide for leachability (migrability) testing for children's products considered in future legislation.

Education:

We greatly appreciate the inclusion of Sec. 5484 regarding resources for educating the general public, retailers, and manufacturers. Recognizing the state's current financial situation we understand the limited capacity for an educational component at this time and hope that additional resources will be devoted to this when the state's finances improve.

Violations:

We appreciate the inclusion of a civil fine on a per item basis and the inclusion of a stiffer fine for those who knowingly violate these regulations.

Extension to Toxic Metal Content:

Unfortunately lead is not the only toxic metal in common use. There are also standards for Antimony, Arsenic, Barium, Cadmium, Chromium, Mercury, and Selenium. For example there are currently voluntary standards for these materials for toys. Several states have passed legislation that mandates the reduction of these metals in packaging materials (please see the documents attached).

We encourage the committee to consider additional legislation that will address toxic metals as a class.

HB 4399

An effort to regulate lead or toxic metals generally in lunch boxes is an important effort. This is a logical addition to the new Part 54B section of the Michigan Public Health Code.

We encourage the committee to continue to expand the number and types of children's products that would be covered under Part 54B.

Thank-you again for your efforts to protect children from toxins in children's products and for your consideration of our comments.



Toxics in Packaging Clearinghouse

c/o Northeast Recycling Council, Inc. (NERC)

139 Main Street, Suite 401 ■ Brattleboro, VT 05301

802.254.8911 ■ www.toxicsinpackaging.org

FACT SHEET

Introduction

The Model Toxics in Packaging Legislation was developed in 1989 to reduce the amount of four heavy metals in packaging and packaging components sold or distributed throughout the states. As of July 2004, legislation based on this model has been adopted by nineteen states:

- | | | |
|---------------|-----------------|----------------|
| • California | • Maryland | • Rhode Island |
| • Connecticut | • Minnesota | • Vermont |
| • Florida | • Missouri | • Virginia |
| • Georgia | • New Hampshire | • Washington |
| • Illinois | • New Jersey | • Wisconsin |
| • Iowa | • New York | |
| • Maine | • Pennsylvania | |

The influence of the Model Legislation extends beyond US borders. The European Union, for example, used the Model as the basis of its packaging requirements (94/62/EC).

Incidental Presence Concentration Limits

No intentional introduction of any amount of the four metals is allowed. The sum of the concentration levels of incidentally introduced lead, mercury, cadmium, and hexavalent chromium present in any package or packaging component shall not exceed the following:

- 600 parts per million, two years after enactment
- 250 parts per million, three years after enactment
- 100 parts per million, four years after enactment

Who is Responsible?

- Manufacturers of packaging and packaging components
- Suppliers of packaging and packaging components
- Product manufacturers or distributors who use packaging

How to Comply

The manufacturer or supplier to the purchaser must submit a certificate of compliance stating that a package or packaging component is in compliance with the requirements of the law. (This provision does not apply to the individual making retail purchases or to retail storeowners.) The purchaser, manufacturer and supplier should keep a copy of the signed certificate of compliance on file as long as that package is in use. The certificate of compliance can be subject to state and public review upon request.

Enforcement

Enforcement of the Model Toxics in Packaging Legislation is at the discretion of each individual state. However, violation information will be shared among the Clearinghouse member states, and will be pursued in a consistent manner, to the extent possible.

Exemptions

Details of these exemptions can be found in the individual state laws, and specific exemptions may vary by state. All packages and packaging components are subject to the law except:

- Packages and packaging components with a code indicating that the date of manufacture was prior to the effective date of the law.
- Packages and packaging components to which heavy metals have been added in order to comply with health and safety requirements specified by federal law. (2-year exemption—requires approval)
- Packages and packaging components that would not exceed the maximum contaminant levels, but for the addition of recycled materials. This exemption does not apply to use of the metals when they have already been recovered and separated for use as a metal or metallic compound. (Expires Jan. 1, 2010)
- Packages and packaging components to which heavy metals have been added in the manufacturing process for which there is no feasible or technical alternative. (2-year exemption—requires approval)
- Packages and packaging components that exceed the contaminant levels, but are reused; and the enclosed product, its transportation and disposal are regulated by federal health and safety requirements. (Expires Jan. 1, 2010)
- Packages and packaging components that exceed the contaminant levels but have a controlled distribution and are reused. (Expires Jan. 1, 2010—requires approval)
- A glass package or packaging component that has a vitrified label.

More Information Online

See www.toxicsinpackaging.org, which includes:

- 2004 revised model legislation
- Q&A document, which lists the most commonly asked questions regarding the toxics in packaging legislation
- Sample certificate of compliance and certificate of exemption
- Comparative Analysis, presenting a side-by-side comparison of the model legislation and existing state laws.

Interested in Joining?

Membership categories include:

- States that have enacted toxics in packaging legislation
- States considering adoption of the legislation
- Industry/Trade Associations
- Non-Profit Organizations

Revised January, 2005

For more information about the TPCH, please contact the Northeast Recycling Council, Inc. (NERC)

BILL TEXT:

STATE OF NEW YORK

5784

2007-2008 Regular Sessions

IN SENATE

May 9, 2007

Introduced by Sens. ALESI, RATH, TRUNZO -- read twice and ordered printed, and when printed to be committed to the Committee on Environmental Conservation

AN ACT to amend the environmental conservation law, in relation to jewelry containing lead

The People of the State of New York, represented in Senate and Assembly, do enact as follows:

- 1 Section 1. Legislative findings. The legislature hereby finds that
- 2 stringent controls on the amount of lead in jewelry are necessary to
- 3 protect public health, especially the health of children. Random
- 4 samples of jewelry in New York state have been found to contain up to
- 5 60,000 parts per million of lead. To assure consistent application of
- 6 these controls to all jewelry, specific technical standards and controls
- 7 must be specified.
- 8 § 2. The environmental conservation law is amended by adding a new
- 9 section 37-0113 to read as follows:
- 10 § 37-0113. Lead-containing jewelry.
- 11 For purposes of this section, the following terms shall have the
- 12 following definitions: 1. "Body piercing jewelry" means any part of
- 13 jewelry that is manufactured or sold for placement in a new piercing or
- 14 a mucous membrane, but does not include any part of that jewelry that is
- 15 not placed within a new piercing or a mucous membrane.
- 16 2. "Children" means children aged six and younger.
- 17 3. "Children's jewelry" means jewelry that is made for, marketed for
- 18 use by, or marketed to, children. Children's jewelry includes, but is
- 19 not limited to, jewelry that meets any of the following conditions:
- 20 (a) Represented in its packaging, display, or advertising, as appro-

21 piate for use by children.

22 (b) Sold in conjunction with, attached to, or packaged together with
23 other products that are packaged, displayed, or advertised as appropri-
24 ate for use by children.

EXPLANATION--Matter in *italics* (underscored) is new; matter in brackets
[] is old law to be omitted.

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1 (c) Sized for children and not intended for use by adults.

2 (d) Sold in any of the following:

3 (1) A vending machine.

4 (2) Retail store, catalogue, or online web site, in which a person
5 exclusively offers for sale products that are packaged, displayed, or
6 advertised as appropriate for use by children.

7 (3) A discrete portion of a retail store, catalogue, or online web
8 site, in which a person offers for sale products that are packaged,
9 displayed, or advertised as appropriate for use by children.

10 4. "Class 1 material" means any of the following materials:

11 (a) stainless or surgical steel;

12 (b) karat gold;

13 (c) sterling silver;

14 (d) platinum, palladium, iridium, ruthenium, rhodium or osmium;

15 (e) natural or cultured pearls;

16 (f) glass, ceramic, or crystal decorative components, including cat's
17 eye, cubic zirconia, including cubic zirconium or cz, rhinestones, and
18 cloisonne;

19 (g) a gemstone that is cut and polished for ornamental purposes;

20 (h) elastic, fabric, ribbon, rope, or string, unless it contains
21 intentionally added lead and is listed as a class 2 material;

22 (i) all natural decorative material, including amber, bone, coral,
23 feathers, fur, horn, leather, shell, wood, that is in its natural state
24 and is not treated in a way that adds lead; and

25 (j) adhesive.

26 (k) The following gemstones are not class 1 materials: aragonite,
27 bayldonite, boleite, cerussite, crocoite, ekanite, linarite, mimetite,
28 phosgenite, samarskite, vanadinite, and wulfenite.

29 5. "Class 2 material" means any of the following materials:

30 (a) electroplated metal that meets the following standards:

31 (1) on and before August 30, 2009, a metal alloy with less than ten
32 percent lead by weight that is electropolated with suitable under and
33 finish coats.

34 (2) on and after August 31, 2009, a metal alloy with less than six
35 percent lead by weight that is electroplated with suitable under and
36 finish coats; or
37 (b) unplated metal with less than 1.5 percent lead that is not other-
38 wise listed as a class 1 material; or
39 (c) plastic or rubber, including acrylic, polystyrene, plastic beads
40 and stones, and polyvinyl chloride (PVC) that meets the following stand-
41 ards:
42 (1) on and before August 30, 2009, less than 0.06 percent (six hundred
43 parts per million) lead by weight; and
44 (2) on and after August 31, 2009, less than 0.02 percent (two hundred
45 parts per million) lead by weight; or
46 (d) a dye or surface coating containing less than 0.06 percent (six
47 hundred parts per million) lead by weight.
48 6. "Class 3 material" means any portion of jewelry that meets both of
49 the following criteria:
50 (a) is not a class 1 or class 2 material; and
51 (b) contains less than 0.06 percent (six hundred parts per million)
52 lead by weight.
53 7. "Component" means any part of jewelry.
54 8. "EPA reference methods 3050B (acid digestion of sediments, sludges
55 and soils) or 3051 (microwave assisted digestion/sludge, soils)" means
56 those test methods incorporated by reference in paragraph eleven of

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1 subdivision (a) of section 260.11 of title 40 of the code of federal
2 regulations.
3 9. "Jewelry" means:
4 (a) any of the following ornaments worn by a person: an anklet, arm
5 cuff, bracelet, brooch, chain, crown, cuff link, decorated hair accesso-
6 ries, earring, necklace, pin, ring, or body piercing jewelry; or
7 (b) any bead, chain, link, pendant, or other component of such an
8 ornament.
9 10. (a) "Surface coating" means a fluid, semifluid, or other material,
10 with or without a suspension of finely divided coloring matter, that
11 changes to a solid film when a thin layer is applied to a metal, wood,
12 stone, paper, leather, cloth, plastic, or other surface.
13 (b) "Surface coating" does not include a printing ink or a material
14 that actually becomes a part of the substrate, including, but not limit-
15 ed to, pigment in a plastic article, or a material that is actually
16 bonded to the substrate, such as by electroplating or ceramic glazing.
17 11. On or after March 1, 2008, no person shall advertise, manufacture,
18 offer for sale, sell, or distribute for promotional purposes in this

19 state, or import for distribution or sale in this state, any jewelry
 20 unless the jewelry is made entirely from a class 1, class 2, or class 3
 21 material, or any combination thereof.

22 12. Notwithstanding subdivision ten of this section, on or after
 23 September 1, 2007, no person shall advertise, manufacture, offer for
 24 sale, sell, or distribute for promotional purposes in this state, or
 25 import for distribution or sale in this state, any children's jewelry
 26 unless the children's jewelry is made entirely from one or more of the
 27 following materials:

- 28 (a) a nonmetallic material that is a class 1 material;
- 29 (b) a nonmetallic material that is a class 2 material;
- 30 (c) a metallic material that is either a class 1 material or contains
 31 less than 0.06 percent (six hundred parts per million) lead by weight;
- 32 (d) glass or crystal decorative components that weigh in total no more
 33 than one gram, excluding any glass or crystal decorative component that
 34 contains less than 0.02 percent (two hundred parts per million) lead by
 35 weight and has no intentionally added lead;
- 36 (e) printing ink or ceramic glaze that contains less than 0.06 percent
 37 (six hundred parts per million) lead by weight; or
- 38 (f) class 3 material that contains less than 0.02 percent (two hundred
 39 parts per million) lead by weight.

40 13. Notwithstanding subdivision ten of this section, on or after March
 41 1, 2008, no person shall advertise, manufacture, offer for sale, sell,
 42 or distribute for promotional purposes in this state, or import for
 43 distribution or sale in this state, any body piercing jewelry unless the
 44 body piercing jewelry is made of one or more of the following materials:
 45 surgical implant stainless steel, surgical implant grade of titanium,
 46 niobium (NB), solid fourteen karat or higher white or yellow nickel-free
 47 gold, solid platinum, or a dense low-porosity plastic, including, but
 48 not limited to, tygon or polytetrafluoroethylene (PTFE), if the plastic
 49 contains no intentionally added lead.

50 § 3. The environmental conservation law is amended by adding a new
 51 section 37-0115 to read as follows:

52 § 37-0115. Testing methods for determining compliance with section
 53 37-0113.

54 1. The testing methods for determining compliance with section 37-0113
 55 of this title shall be conducted using the EPA reference methods 3050B
 56 or 3051 for the material being tested, except as otherwise provided in

1 subparagraphs 5 and 6 of paragraph (e) of subdivision 2 of this section,
 2 and in accordance with all of the following procedures:

- 3 (a) When preparing a sample, the laboratory shall make every effort to

4 assure that the sample removed from a jewelry piece is representative of
5 the component to be tested, and is free of contamination from extraneous
6 dirt and material not related to the jewelry component to be tested.

7 (b) All jewelry component samples shall be washed prior to testing
8 using standard laboratory detergent, rinsed with laboratory reagent
9 grade deionized water, and dried in a clean ambient environment.

10 (c) If a component is required to be cut or scraped to obtain a
11 sample, the metal snips, scissors, or other cutting tools used for the
12 cutting or scraping shall be made of stainless steel and washed and
13 rinsed before each use and between samples.

14 (d) A sample shall be digested in a container that is known to be free
15 of lead and with the use of an acid that is not contaminated by lead,
16 including analytical reagent grade digestion acids and reagent grade
17 deionized water.

18 (e) Method blanks, consisting of all reagents used in sample prepara-
19 tion handled, digested, and made to volume in the same exact manner and
20 in the same container type as samples, shall be tested with each group
21 of twenty or fewer samples tested.

22 (f) The results for the method blanks shall be reported with each
23 group of sample results, and shall be below the stated reporting limit
24 for sample results to be considered valid.

25 2. In addition to the requirements of subdivision one of this section,
26 the following procedures shall be used for testing the following mate-
27 rials:

28 (a) For testing a metal plated with suitable undercoats and finish
29 coats, the following protocols shall be observed:

30 (1) Digestion shall be conducted using hot concentrated nitric acid
31 with the option of using hydrochloric acid or hydrogen peroxide.

32 (2) The sample size shall be 0.050 gram to one gram.

33 (3) The digested sample may require dilution prior to analysis.

34 (4) The digestion and analysis shall achieve a reported detection
35 limit no greater than 0.1 percent for samples.

36 (5) All necessary dilutions shall be made to ensure that measurements
37 are made within the calibrated range of the analytical instrument.

38 (b) For testing unplated metal and metal substrates that are not a
39 class 1 material the following protocols shall be observed:

40 (1) Digestion shall be conducted using hot concentrated nitric acid
41 with the option of using hydrochloric acid and hydrogen peroxide.

42 (2) The sample size shall be 0.050 gram to one gram.

43 (3) The digested sample may require dilution prior to analysis.

44 (4) The digestion and analysis shall achieve a reported detection
45 limit no greater than 0.01 percent for samples.

46 (5) All necessary dilutions shall be made to ensure that measurements
47 are made within the calibrated range of the analytical instrument.

48 (c) For testing polyvinyl chloride (PVC), the following protocols
49 shall be observed:

50 (1) The digestion shall be conducted using hot concentrated nitric
51 acid with the option of using hydrochloric acid and hydrogen peroxide.
52 (2) The sample size shall be a minimum of 0.05 gram if using microwave
53 digestion or 0.5 gram if using hotplate digestion, and shall be chopped
54 or comminuted prior to digestion.
55 (3) Digested samples may require dilution prior to analysis.

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1 (4) Digestion and analysis shall achieve a reported detection limit no
2 greater than 0.001 percent (ten parts per million) for samples.
3 (5) All necessary dilutions shall be made to ensure that measurements
4 are made within the calibrated range of the analytical instrument.
5 (d) For testing plastic or rubber that is not polyvinyl chloride
6 (PVC), including acrylic, polystyrene, plastic beads, or plastic stones,
7 the following protocols shall be observed:
8 (1) The digestion shall be conducted using hot concentrated nitric
9 acid with the option of using hydrochloric acid or hydrogen peroxide.
10 (2) The sample size shall be a minimum of 0.05 gram if using microwave
11 digestion or 0.5 gram if using hotplate digestion, and shall be chopped
12 or comminuted prior to digestion.
13 (3) Plastic beads or stones shall be crushed prior to digestion.
14 (4) Digested samples may require dilution prior to analysis.
15 (5) Digestion and analysis shall achieve a reported detection limit no
16 greater than 0.001 percent (ten parts per million) for samples.
17 (6) All necessary dilutions shall be made to ensure that measurements
18 are made within the calibrated range of the analytical instrument.
19 (e) For testing coatings on glass and plastic pearls, the following
20 protocols shall be observed:
21 (1) The coating of glass or plastic beads shall be scraped onto a
22 surface free of dust, including a clean weighing paper or pan, using a
23 clean stainless steel razor blade or other clean sharp instrument that
24 will not contaminate the sample with lead. The substrate pearl material
25 shall not be included in the scrapings.
26 (2) The razor blade or sharp instrument shall be rinsed with deionized
27 water, wiped to remove particulate matter, rinsed again, and dried
28 between samples.
29 (3) The scrapings shall be weighed and not less than fifty micrograms
30 of scraped coating shall be used for analysis. If less than fifty micro-
31 grams of scraped coating is obtained from an individual pearl, multiple
32 pearls from that sample shall be scraped and composited to obtain a
33 sufficient sample amount.
34 (4) The number of pearls used to make the composite shall be noted.
35 (5) The scrapings shall be digested according to EPA reference method

36 3050B or 3051 or an equivalent procedure for hot acid digestion in prep-
37 aration for trace lead analysis.
38 (6) The digestate shall be diluted in the minimum volume practical for
39 analysis.
40 (7) The digested sample shall be analyzed according to specification
41 of an approved and validated methodology for inductively coupled plasma
42 mass spectrometry.
43 (8) A reporting limit of 0.001 percent (ten parts per million) in the
44 coating shall be obtained for the analysis.
45 (9) The sample result shall be reported within the calibrated range of
46 the instrument. If the initial test of the sample is above the highest
47 calibration standard, the sample shall be diluted and reanalyzed within
48 the calibrated range of the instrument.
49 (f) For testing dyes, paints, coatings, varnish, printing inks, ceram-
50 ic glazes, glass, or crystal, the following testing protocols shall be
51 observed:
52 (1) The digestion shall use hot concentrated nitric acid with the
53 option of using hydrochloric acid or hydrogen peroxide.
54 (2) The sample size shall be not less than 0.050 gram, and shall be
55 chopped or comminuted prior to digestion.
56 (3) The digested sample may require dilution prior to analysis.

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1 (4) The digestion and analysis shall achieve a reported detection
2 limit no greater than 0.001 percent (ten parts per million) for samples.
3 (5) All necessary dilutions shall be made to ensure that measurements
4 are made within the calibrated range of the analytical instrument.
5 (g) For testing glass and crystal used in children's jewelry, the
6 following testing protocols for determining weight shall be used:
7 (1) A component shall be free of any extraneous material, including
8 adhesive, before it is weighed.
9 (2) The scale used to weigh a component shall be calibrated immediate-
10 ly before the components are weighed using s-class weights of one and
11 two grams, as certified by the National Institute of Standards and Tech-
12 nology (NIST) of the Department of Commerce.
13 (3) The calibration of the scale shall be accurate to within 0.01
14 gram.
15 3. The commissioner may promulgate rules and/or regulations modifying
16 the testing protocols specified in subdivisions one and two of this
17 section, as such commissioner deems necessary to further the purposes of
18 this section.
19 § 4. The environmental conservation law is amended by adding a new
20 section 71-3711 to read as follows:

21 § 71-3711. Enforcement of section 37-0113.

22 1. Any person who knowingly or intentionally violates any provision of
23 or fails to perform any duty imposed by section 37-0113 of this chapter
24 shall upon the first finding of such a violation be liable for a civil
25 penalty not to exceed one hundred dollars for each violation. Any
26 person convicted of a second or subsequent violation shall be liable for
27 a civil penalty not to exceed twenty-five hundred dollars for each
28 violation.

29 2. Penalties under this section shall be assessed by the commissioner
30 after a hearing pursuant to the provisions of section 71-1709 of this
31 article, and in addition thereto, any person found to have violated
32 section 37-0113 of this chapter may be enjoined from continuing such
33 violation.

34 3. All civil penalties and fines collected for any violation of
35 section 37-0113 of this chapter shall be paid over to the commissioner
36 for deposit in the environmental protection fund established by section
37 ninety-two-s of the state finance law.

38 4. (a) No charge of a violation of the provisions of, or failure to
39 perform a duty imposed by section 37-0113 of this chapter shall be based
40 upon excessive lead content except upon a showing that the laboratory
41 tests establishing such excessive lead content were performed by a labo-
42 ratory that complies with the testing requirements established by
43 section 37-0115 of this chapter.

44 (b) A person charged with a violation of the provisions of, or failure
45 to perform a duty imposed by section 37-0113 of this chapter shall be
46 provided with all supporting documentation related to the testing of the
47 jewelry, including, but not limited to, documentation of the procedures
48 utilized by the laboratory, copies of all test results, exemplars of the
49 products tested to the extent practicable, and such other documentation
50 and evidence which shall reasonably be required to verify the accuracy
51 of the test results.

52 § 5. This act shall take effect immediately.

Michigan Network for Children's Environmental Health

— A Brief Overview

Who We Are

- We are a coalition of health professionals, health-affected groups, environmental organizations, and others dedicated to a safe and less toxic world for Michigan's children.

Our Mission

- Through education, outreach, and advocacy, we seek to protect Michigan's children from adverse impacts caused by exposure to widespread hazardous chemicals.

What We Do

- Advocate for policy changes at the state and local levels to reduce threats to children's health.
- Educate health professionals and the general public.
- Build the case for broad reform of chemicals regulation to protect Michigan's children.

2007 Campaigns

- Protect children's health by urging legislators to:
 - Support legislation to remove hazardous chemicals from articles and toys made for use by children.
 - Phase out pharmaceutical use of the chemical ingredient lindane in lice and scabies treatments.
 - Restrict the manufacture and sale of products containing the toxic, persistent flame retardant deca-BDE.
 - Target products containing the neurotoxic chemical mercury, including barometers, hydrometers, manometers, and relays, and align state purchasing with a mercury-free goal.
- Advance the Governor's Executive Directive on "Green Chemistry:"
 - Green Chemistry is the design of chemical products and processes to reduce or eliminate the use and generation of hazardous substances. Green Chemistry is an innovative approach to protecting children's health, reducing environmental damage, and increasing the economic competitiveness of businesses in Michigan.



Michigan Network for
**CHILDREN'S
Environmental
HEALTH**

MEMBER ORGANIZATIONS

- American Academy of Pediatrics (Michigan Chapter)
- Arab Community Center for Economic and Social Services (ACCESS)
- Association for Children's Mental Health
- Clean Water Fund
- Clinton County Family Resource Center
- East Michigan Environmental Action Council (EMEAC)
- Ecology Center
- Healthy Homes Coalition of West Michigan
- Learning Disabilities Association (LDA) of Michigan
- Local Motion
- Michigan Council for Maternal and Child Health
- Michigan Environmental Council
- Science and Environmental Health Network

JOIN THE NETWORK

For more information or to join the Network:

- E-mail: healthykids@mnceh.org
Tel: 734-761-3186 x115
- Visit our website and sign-up on-line at www.mnceh.org

Developmental neurotoxicity of industrial chemicals



P Grandjean, PJ Landrigan

Neurodevelopmental disorders such as autism, attention deficit disorder, mental retardation, and cerebral palsy, and autism are common, costly, and can cause lifelong disability. Their causes are mostly unknown. A few industrial chemicals (eg, lead, methylmercury, polychlorinated biphenyls [PCBs], arsenic, and toluene) are recognised causes of neurodevelopmental disorders and subclinical brain dysfunction. Exposure to these chemicals during early fetal development can cause brain injury at doses much lower than those affecting adult brain function. Recognition of these risks has led to evidence-based programmes of prevention, such as elimination of lead additives in petrol. Although these prevention campaigns are highly successful, most were initiated only after substantial delays. Another 200 chemicals are known to cause clinical neurotoxic effects in adults. Despite an absence of systematic testing, many additional chemicals have been shown to be neurotoxic in laboratory models. The toxic effects of such chemicals in the developing human brain are not known and they are not regulated to protect children. The two main impediments to prevention of neurodevelopmental deficits of chemical origin are the great gaps in testing chemicals for developmental neurotoxicity and the high level of proof required for regulation. New, precautionary approaches that recognise the unique vulnerability of the developing brain are needed for testing and control of chemicals.

One in every six children has a developmental disability and in most cases these disabilities affect the nervous system.¹ The most common neurodevelopmental disorders include learning disabilities, sensory deficits, developmental delays, and cerebral palsy.¹ Some experts have reported that the prevalence of certain neurodevelopmental disorders—autism and attention deficit and hyperactivity disorder, in particular—might be increasing, but there are few data to sustain that position.² Treatment of these disorders is difficult, and the disabilities they cause can be permanent;³ they are therefore very costly to families and to society.⁴⁻⁶

Evidence has been accumulating over several decades that industrial chemicals can cause neurodevelopmental damage and that subclinical stages of these disorders might be common. The possibility of a link between chemicals and widespread neurobehavioural changes was first raised by research showing that lead was toxic to the developing brain across a wide range of exposures.⁷⁻¹⁰ That report was in accord with reports indicating that other environmental pollutants were also toxic to early brain development.¹¹ An expert committee from the US National Research Council concluded that 3% of developmental disabilities are the direct result of environmental exposure to such poisons, and that another 25% arise through interactions between environmental factors and individual genetic susceptibility.¹ These estimates were based on scarce information about neurotoxicity and could therefore underestimate the true prevalence of chemically-induced abnormalities.

Neurobehavioural damage caused by industrial chemicals is, in theory, preventable. An essential prerequisite to prevention is recognition of a chemical's ability to harm the developing brain. Knowledge that a chemical is neurotoxic can prompt efforts to restrict its use and to control exposure. Previous evidence-based programmes of exposure prevention, such as those directed against children's exposure to lead, have been

highly successful, although they were initiated after substantial delay.

The aim of this review is to characterise the vulnerability of the developing nervous system to chemical toxicity; to collate publicly available data for human neurotoxicity of industrial chemicals; to examine the possible extent of a developmental neurotoxicity pandemic; to describe the known consequences of developmental neurotoxicity for individuals and society; to examine the implications for human health of the dearth of toxicological information; and to consider prospects for prevention of exposure.

Vulnerability of the developing brain

The developing human brain is inherently much more susceptible to injury caused by toxic agents than is the brain of an adult.¹² This susceptibility stems from the fact that during the 9 months of prenatal life, the human brain must develop from a strip of cells along the dorsal ectoderm of the fetus into a complex organ consisting of billions of precisely located, highly interconnected, and specialised cells. Optimum brain development requires that neurons move along precise

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Search strategy and selection criteria

We identified industrial chemicals that have caused neurotoxic effects in man from the hazardous substances data bank of the US National Library of Medicine, supplemented by fact sheets by the US Agency for Toxic Substances and Disease Registry, and the integrated risk information system of the US Environmental Protection Agency. We searched for the terms "neurotoxic", "neurological", and "neuro". For all neurotoxic substances identified, we then used synonyms, commercial names, and CAS (chemical abstracts service) numbers to search PubMed, TOXNET, and TOXLINE to identify published data for developmental neurotoxicity. The primary search terms were "prenatal exposure delayed effects" [MeSH] and "neurotoxicity syndromes" [MeSH]. Secondary searches used combinations of "maternal exposure" and "maternal fetal exchange" with "developmental disabilities/chemically induced" and "neurotoxins", all with the limiters "all child: 0-18 years", "most recent 10 Years", "English", and "human". We also used references cited in the chosen articles.

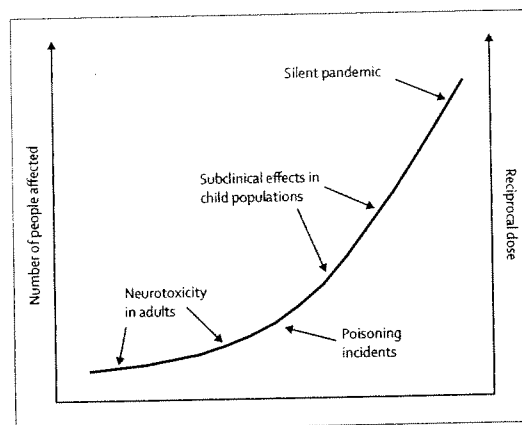


Figure 1: The effects of a neurotoxic chemical on a population over time. For identification of chemicals toxic to neurodevelopment, the first evidence dealt with adverse effects of high doses on the adult nervous system, and was followed by case reports and epidemiological evidence on developmental toxicity at successively lower doses, to which childhood populations of increasing magnitude were exposed. Recognition of inorganic lead, methylmercury, and polychlorinated biphenyls as neurotoxic followed this curve towards the right, and arsenic and toluene were later seen to match this curve. Documentation of most neurotoxins is directed toward adults only and therefore many compounds remain far to the left on the timescale.

pathways from their points of origin to their assigned locations, that they establish connections with other cells, both nearby and distant, and that they learn to communicate with other cells via such connections.¹²⁻¹⁴ All these processes have to take place within a tightly controlled time frame, in which each developmental stage has to be reached on schedule and in the correct sequence. Because of the extraordinary complexity of human brain development, windows of unique susceptibility to toxic interference arise that have no counterpart in the mature brain, or in any other organ. If a developmental process in the brain is halted or inhibited, there is little potential for later repair, and the consequences can therefore be permanent.^{12,14}

During fetal development, the placenta offers some protection against unwanted chemical exposures, but it is not an effective barrier against environmental pollutants.¹⁵ For example, many metals easily cross the placenta, and the mercury concentration in umbilical cord blood can be substantially higher than in maternal blood.¹⁶ The blood-brain barrier, which protects the adult brain from many toxic chemicals, is not completely formed until about 6 months after birth.¹⁷

The human brain continues to develop postnatally, and the period of heightened vulnerability therefore extends over many months, through infancy and into early childhood. Although most neurons have been formed by the time of birth, growth of glial cells and myelination of axons continues for several years.^{13,14}

The susceptibility of infants and children to industrial chemicals is further enhanced by their increased exposures, augmented absorption rates, and diminished

ability to detoxify many exogenous compounds, relative to that of adults.^{18,19} Persistent lipophilic substances, including specific pesticides and halogenated industrial compounds, such as PCBs, accumulate in maternal adipose tissue and are passed on to the infant via breast milk, resulting in infant exposure that exceeds the mother's own exposure by 100-fold on the basis of bodyweight.²⁰

Recognition of neurotoxicity

Developmental neurotoxicity in children exposed to industrial chemicals is often first identified through recognition of obvious functional abnormalities after high-dose exposure that clearly caused poisoning. Good quality research later documented the presence of less striking, but nonetheless serious adverse effects at low doses of exposure (figure 1). This sequence of discovery led to the recognition that environmental poisons exert a range of adverse effects—some are clinically evident, but others can be discerned only through special testing and are not evident on standard examination, hence the term subclinical toxicity. The underlying idea is that there is a dose-dependent continuum of toxic effects, in which clinically obvious effects have subclinical counterparts.²¹ A pandemic of subclinical neurotoxicity is therefore likely to be silent—ie, not apparent from standard health statistics.

The notion of subclinical toxicity originates from Landrigan⁷ and Needleman⁸ and their colleagues' pioneering work who showed that children's exposure to lead could cause reductions in intelligence and changes in behaviour even in the absence of clinically visible symptoms of lead toxicity. The subclinical toxicity of lead in children has subsequently been confirmed in prospective epidemiological studies.^{22,23}

Parallel findings have been reported on some other industrial chemicals, but their number is small. About 80 000 chemicals are registered for commercial use with the US Environmental Protection Agency, and 62 000 were already in use when the Toxic Substances Control Act was enacted in the USA in 1977.²⁴ The situation is similar in the EU, where 100 000 chemicals were registered in 1981.²⁵ The full extent to which these chemicals contribute to neurodevelopmental disorders and subclinical neurotoxicity is still unknown.

Neurotoxic agents Identification

Studies in animals support the notion that a wide range of industrial chemicals can cause developmental neurotoxicity at low doses that are not harmful to mature organisms.^{26,27} Such injury seems to result in permanent changes in brain function that might become detectable only when the animal reaches maturity. Because developmental neurotoxicity might not be apparent from routine toxicology tests,²⁸ identification of neurotoxic chemicals often rests on clinical and epidemiological data.

To identify environmental chemicals that are toxic to the human brain, we searched the hazardous substances data bank of the US National Library of Medicine, where substances are listed with their adverse effects in man. We checked the completeness of this list against other data sources and with a previous review of published data for clinical toxicity.²⁹ The panel shows the industrial chemicals known to be neurotoxic in man. We have excluded drugs, food additives, microbial toxins, and snake venoms and similar biogenic substances. This list excludes chemicals that have proved neurotoxic solely in laboratory animals, for which no systematic list exists. We mainly include acute poisons that have caused serious accidents or have been used in suicide attempts, Neurotoxins that mainly cause chronic or delayed disease are likely to be underrepresented.²⁹ The largest groups of identified compounds are metals, solvents, and pesticides, but other chemicals with less documentation could have unrecognised effects. The list therefore should not be regarded as comprehensive.

Many substance names (see panel) were used for searches of published data for developmental neurotoxicity. On the basis of our critical review, the few known chemicals causing neurodevelopmental abnormalities are highlighted the panel. Many more chemicals that we have not listed are known to be neurodevelopment poisons in laboratory animals,²⁷ but no data about their potential toxic effects on human brain development are available.

Lead

The neurotoxic effects of lead in adults were known in Roman times, but a report from Australia 100 years ago was the first description of epidemic lead poisoning in young children; the source of the outbreak was traced to ingestion of lead-based paint by children playing on verandas with peeling paint.³⁰ Further reports of childhood lead poisoning from the USA and Europe followed. Lead poisoning was at that time thought to be an acute illness, from which a child either recovered or died. Long-term sequelae were first documented in the 1940s, when 19 of 20 survivors of acute poisoning were noted to have severe learning and behavioural problems.³¹

Despite those early paediatric warnings, the largely unchecked use of lead in petrol, paints, ceramic glazes, and many other products through much of the twentieth century caused continued risk of lead poisoning. During the 1970s, widespread subclinical neurobehavioural deficits, including problems with concentration, memory, cognition, and behaviour, were documented in asymptomatic children with raised blood-lead concentrations.^{7,8} Spurred by recommendations issued by the European Regional Office of WHO, studies were initiated in many countries; the results corroborated the previous conclusions.³²

As a result of accumulating evidence, many sources of lead exposure became controlled, although not all sources,

Panel: Chemicals (n=201) known to be neurotoxic in man

Metals and inorganic compounds

- Aluminum compounds
- *Arsenic and arsenic compounds
- Azide compounds
- Barium compounds
- Bismuth compounds
- Carbon monoxide
- Cyanide compounds
- Decaborane
- Diborane
- Ethylmercury
- Fluoride compounds
- Hydrogen sulphide
- *Lead and lead compounds
- Lithium compounds
- Manganese and manganese compounds
- Mercury and mercuric compounds
- *Methylmercury
- Nickel carbonyl
- Pentaborane
- Phosphine
- Phosphorus
- Selenium compounds
- Tellurium compounds
- Thallium compounds
- Tin compounds

Organic solvents

- Acetone
- Benzene
- Benzyl alcohol
- Carbon disulphide
- Chloroform
- Chloroprene
- Cumene
- Cyclohexane
- Cyclohexanol
- Cyclohexanone
- Dibromochloropropane
- Dichloroacetic acid
- 1,3-Dichloropropene
- Diethylene glycol
- N,N-Dimethylformamide
- 2-Ethoxyethyl acetate
- Ethyl acetate
- Ethylene dibromide
- Ethylene glycol
- n-Hexane
- Isobutyronitrile
- Isophorone
- Isopropyl alcohol

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- Isopropylacetone

- Methanol
 - Methyl butyl ketone
 - Methyl cellosolve
 - Methyl ethyl ketone
 - Methylcyclopentane
 - Methylene chloride
 - Nitrobenzene
 - 2-Nitropropane
 - 1-Pentanol
 - Propyl bromide
 - Pyridine
 - Styrene
 - Tetrachloroethane
 - Tetrachloroethylene
 - *Toluene
 - 1,1,1-Trichloroethane
 - Trichloroethylene
 - Vinyl chloride
 - Xylene
- Other organic substances**
- Acetone cyanohydrin
 - Acrylamide
 - Acrylonitrile
 - Allyl chloride
 - Aniline
 - 1,2-Benzenedicarbonitrile
 - Benzonitrile
 - Butylated triphenyl phosphate
 - Caprolactam
 - Cydonite
 - Dibutyl phthalate
 - 3-(Dimethylamino)-propanenitrile
 - Diethylene glycol diacrylate
 - Dimethyl sulphate
 - Dimethylhydrazine
 - Dinitrobenzene
 - Dinitrotoluene
 - Ethylbis(2-chloroethyl)amine
 - Ethylene
 - Ethylene oxide
 - Fluoroacetamide
 - Fluoroacetic acid
 - Hexachlorophene
 - Hydrazine
 - Hydroquinone
 - Methyl chloride
 - Methyl formate
 - Methyl iodide
 - Methyl methacrylate
 - p-Nitroaniline
 - Phenol

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- p-Phenylenediamine

and not in all countries. A 90% reduction in childhood blood-lead concentrations followed the termination of lead additives in petrol.³³ Now, research into lead neurotoxicity focuses on the shape of the dose-response curve at very low exposures that seem to cause surprisingly large functional decrements.³² As convincing evidence was recognised, health agencies reduced the permissible concentration of lead in children's blood. However, up-to-date research³² suggests that the current effects of lead exposure on human brain development could be even greater than previously thought.

Methylmercury

Toxic effects on the brain due to methylmercury were first established in men with occupational exposure.³⁴ The developmental toxicity of this organic mercury compound became evident in the 1960s in Minamata, Japan, where an epidemic of spasticity, blindness and profound mental retardation was seen in infants born to mothers who consumed fish from contaminated waters. After many years of clinical and experimental studies, the source proved to be mercury compounds released into Minamata Bay by a plastics plant.³⁵ Methylmercury accumulated and reached high concentrations in locally caught fish. Exposed adults, including mothers of poisoned children, were less seriously affected, if at all.³⁶ Similar outbreaks of profound neurodevelopmental disorders in the infants of seemingly unaffected mothers have arisen after maternal consumption during pregnancy of seed grain treated with methylmercury fungicides.^{37,38} Studies of a serious poisoning incident in Iraq established a crude dose-response association between mercury concentrations in maternal hair and risk of neurological abnormalities in the children of the women.³⁹

Recent studies have focused on prenatal exposures to reduced concentrations of methylmercury. They have examined populations with a high intake of seafood and freshwater fish with various degrees of methylmercury contamination. Prospective examination of a New Zealand cohort noted a three-point decrement in intelligence quotient (IQ) and changes in affect in children born to women with mercury concentrations in hair of greater than 6 µg/g.⁴⁰ A large prospective study in the Faroe Islands noted evidence of dose-related impairments in memory, attention, language, and visuospatial perception in exposed children.⁴¹ A third prospective cohort study in the Seychelles provided some support for prenatal neurotoxicity after adjustment for postnatal exposures.⁴² Several cross-sectional studies recorded significant associations between methylmercury exposure and neurobehavioral impairment in young children.⁴³

The US National Academy of Sciences reviewed these studies and concluded that strong evidence exists for fetal neurotoxicity of methylmercury, even at low exposures.⁴⁴ These findings have led food safety

authorities to issue dietary advisories, and national and international agencies (with coordination from the UN Environment Programme) now seek to control and restrict mercury releases to the environment. Substantial reductions have already been achieved in mercury use and release from hospitals and incinerators.⁴⁵ A related substance, ethylmercury, has been widely used as a preservative in vaccines, but neurotoxic risk has not been documented.⁴⁶

Arsenic

Ingestion of arsenic-contaminated drinking water has long been recognised to cause peripheral neuropathy in adults.⁴⁷ Developmental neurotoxicity due to arsenic was reported in 1955 in Japan, where consumption of powdered milk contaminated with arsenic led to over 12 000 cases of poisoning and 131 deaths.⁴⁸ A follow-up study of three groups of adolescents born during the time of the milk contamination included one group that was fully breast-fed, one that was exposed to the tainted milk product, and one that received other supplements, but no tainted formula.⁴⁹ Compared with national rates, a ten-fold increase in mentally-retarded individuals were seen in the tainted milk group.⁴⁸ Poor school records, emotional disturbances, and abnormal or borderline electroencephalogram findings were also more common in the exposed group. Since these findings were initially reported in Japanese journals not easily available elsewhere,^{48,49} they have often been overlooked, even in the most thorough risk assessments of environmental arsenic exposure.^{50,51}

Arsenic is present in ground water worldwide, and industrial pollution is widespread. Cross-sectional studies of school-age children showed cognitive deficits associated with drinking water contamination⁵² and raised urinary arsenic concentrations.⁵³ Similar results were obtained in children with arsenic exposure from a smelter.⁵⁴ Possible combined adverse effects on IQ caused by arsenic and manganese exposures was suggested by metal concentrations in hair in children living near a hazardous waste site.⁵⁵ Although evidence for subclinical neurodevelopmental neurotoxicity of arsenic is less well established than for lead and methylmercury, the data are consistent and fit with the high-exposure findings from Japan. Still, regulatory action does not emphasise the need to protect the developing brain against this neurotoxic substance.^{50,51}

Polychlorinated biphenyls

PCBs used to be widely used in electrical equipment as insulators. Human toxicity was first described from industrial exposures,⁵⁶ but neurological effects did not seem important. Developmental toxicity of PCBs was first seen in children exposed to high concentrations in two poisoning events in Asia, where cooking oil had been contaminated by PCBs and related substances during manufacturing. Prenatal exposure in one incident, in

- Phenylhydrazine
- Polybrominated biphenyls
- Polybrominated diphenyl ethers
- *Polychlorinated biphenyls
- Propylene oxide
- TCDD
- Tributyl phosphate
- 2,2',2''-Trichlorotriethylamine
- Trimethyl phosphate
- Tri-o-tolyl phosphate
- Triphenyl phosphate

Pesticides

- Aldicarb
- Aldrin
- Bensulide
- Bromophos
- Carbaryl
- Carbofuran
- Carbophenothion
- α -Chloralose
- Chlordane
- Chlordecone
- Chlorfenvinphos
- Chlormephos
- Chlorpyrifos
- Chlorthion
- Coumaphos
- Cyhalothrin
- Cypermethrin
- 2,4-D
- DDT
- Deltamethrin
- Demeton
- Dialifor
- Diazinon
- Dichlofenthion
- Dichlorvos
- Dieldrin
- Dimefox
- Dimethoate
- Dinitrocresol
- Dinoseb
- Dioxathion
- Disulphoton
- Edifenphos
- Endosulphan
- Endothion
- Endrin
- EPN
- Ethiofencarb
- Ethion
- Ethoprop

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Taiwan, was associated with low birthweight, delayed developmental milestones, and lower IQs in comparison with unexposed siblings.⁵⁷ Exposed boys (but not girls)

- Fenitrothion
- Fensulphothion
- Fenthion
- Fenvalerate
- Fonofos
- Formothion
- Heptachlor
- Heptenophos
- Hexachlorobenzene
- Isobenzan
- Isolan
- Isoxathion
- Leptophos
- Lindane
- Merphos
- Metaldehyde
- Methamidophos
- Methidathion
- Methomyl
- Methyl bromide
- Methyl demeton
- Methyl parathion
- Mevinphos
- Mexacarbate
- Mipafox
- Mirex
- Monocrotophos
- Naled
- Nicotine
- Oxydemeton-methyl
- Parathion
- Pentachlorophenol
- Phorate
- Phosphamidon
- Phospholan
- Propaphos
- Propoxur
- Pyriminil
- Sarin
- Schradan
- Soman
- Sulprofos
- 2,4,5-T
- Tebupirimfos
- Tefluthrin
- Terbufos
- Thiram
- Toxaphene
- Trichlorfon
- Trichloronat

*=substances that have been documented also to cause developmental neurotoxicity.

showed deficits in spatial reasoning. A follow-up study showed growth impairment, slow development, lack of endurance, clumsy movement, and very low IQs.⁵⁸ In a similar incident in Japan, neurological damage seemed less prominent than that of the Taiwan contamination.⁵⁹ Because of the mixed exposures, the specific contribution by PCB to these adverse effects cannot be determined.

Epidemiological studies of asymptomatic populations exposed prenatally to PCBs and related contaminants through maternal diet were done in the USA. Subclinical developmental deficits were shown in the most highly exposed of these children^{60,61} and were associated, at age 11 years, with an average IQ score 6.2 points below that of children with lower exposures.⁶² A Dutch cohort included 418 healthy infants and noted subclinical decrements on neonatal neurological examination and in subsequent developmental tests related to increased PCB exposures.⁶³ Continued follow-up of this cohort suggested that the effects could be modified or masked with age, but were still detectable at age 9 years.⁶⁴ Results from a German cohort were in accord with these findings and also suggested that postnatal PCB exposure from breastfeeding contributes to cognitive deficits.⁶⁵ A possible mechanism through which PCBs injure the developing brain is by interference with maternal thyroid function,^{13,66} which might not harm adult brain functions. Although PCB manufacture has been banned in most nations, and exposures are decreasing, exposures at currently prevalent concentrations could still cause developmental neurotoxicity.⁶⁷

Solvents

Solvent neurotoxicity in adults is well known from acute poisoning cases and from occupational studies.⁶⁸ Ethanol is such a solvent. Intermittent, low-level exposures produce mild inebriating effects, but do not lead to irreversible damage. However, heavy, long-term ethanol intake in adults can lead to serious injury, including Wernicke's syndrome but because such exposures are voluntary, ethanol is not included in the panel.

Fetal alcohol syndrome is qualitatively different from the syndrome in adults. It was originally described in infants of mothers with a serious drinking habit, and involves cognitive and behavioural deficits and changes in facial features. Permanent neurotoxic damage in the mother is not a prerequisite for irreversible effects in the child.⁶⁹ At low consumption, subtle but permanent neurotoxicity, including decreased IQ scores, has been seen.⁷⁰ Effects of alcohol on the fetus could be enhanced by specific genetic polymorphisms.⁷¹

Less reliable documentation is available for other solvents widely used in industry. Because of its anaesthetic effects, toluene has been abused by sniffing, and case reports have reported that infants of mothers who sniffed toluene in pregnancy had abnormally low scores on developmental tests and showed delayed development of speech and motor function.⁷²⁻⁷⁴ Additional evidence of cognitive deficits

in children comes from small studies of mothers who reported occupational exposure to solvents, including toluene, during pregnancy. [A: what were the effects on the children here?]⁷⁵⁻⁷⁷ The women were apparently exposed within permissible workplace limits aimed at prevention of neurotoxicity in the workers themselves. However, these studies do not allow any definite conclusions on the specific hazard and the nature of dose-response associations for developmental neurotoxicity.

Pesticides

More than 600 pesticides are registered, and include insecticides, fungicides, and rodenticides. In the USA alone, about 500 million kg [A: we use SI units is conversion ok?] are applied yearly. Acute pesticide neurotoxicity is well known from occupational exposure studies, poisoning events, and suicide data;⁷⁸ such neurotoxicity is often caused by cholinesterase inhibition by organophosphates.

Developmental neurotoxicity was suggested by an anthropological study of two similar groups of asymptomatic, Yaqui children aged 4–5 years in Mexico.⁷⁸ Those with high exposure to a mix of pesticides, including organophosphates, had diminished short-term memory, hand-eye coordination, and drawing ability, whereas unexposed children of the same tribe showed normal development.⁷⁸ Likewise, preschool children from agricultural communities in the USA showed poorer performance on motor speed and latency than did those of urban communities.⁷⁹ Ecuadorean schoolchildren, whose mothers had been exposed to organophosphates and other pesticides from working in greenhouses during pregnancy, showed visuospatial deficits compared with their unexposed peers.⁸⁰ Current pesticide exposure, measured by urinary excretion of organophosphate metabolites, was associated with delays in the children's simple reaction times.⁸⁰ Acute exposure of American children to the organophosphate pesticide, methyl parathion, was associated with persistent problems in short-term memory and attention span.⁸¹ Prospective epidemiological studies of infants exposed prenatally to the organophosphate, chlorpyrifos, recorded significant decreases in head circumference and birthweight and slowing of reflexes.⁸²⁻⁸⁴ Small head circumference, a risk factor for neurodevelopmental disorders, was seen only in exposed infants who were born to mothers with low expression of PON1, an esterase involved in organophosphate detoxification.⁸³ The effect of chlorpyrifos on bodyweight disappeared after introduction of a ban on residential use.⁸⁴ Although organophosphates can undoubtedly cause developmental neurotoxicity, the data are insufficient to determine the potential hazard to the developing brain posed by individual compounds among the dozens of organophosphates in use worldwide.

Emerging neurotoxic substances

Documentation of developmental effects in human beings for the other compounds listed in the panel is

poor. However, three obvious candidate substances deserve particular attention, including two that have not seemed to cause neurotoxicity in adults.

Manganese

Manganese neurotoxicity in adults has been well documented in occupationally exposed populations; parkinsonism is the classic clinical feature, and subclinical neurotoxicity has also been reported.⁸⁵ Concerns about the developmental neurotoxicity of manganese have emerged because the organic manganese compound methylcyclopentadienyl manganese tricarbonyl has been added to petrol as an antiknock agent in Australia and Canada and could be used in the USA and elsewhere in the future. Manganese can also be present in drinking water. In a prospective study of 247 births in Paris, France,⁸⁶ high manganese concentrations in cord blood were associated with impaired neurobehavioural development, especially on the Brunet-Lezine scales at age 9 months and the McCarthy scales at 3 years. At age 6, no association was seen but only 100 of the original children participated.⁸⁶ Community exposures to manganese released into the environment by combustion of methylcyclopentadienyl manganese tricarbonyl,⁸⁷ exposures from a toxic waste site in the USA,⁸⁸ and from contaminated drinking water in Bangladesh⁸⁹ have been associated with subclinical neurological impairment in children.

Fluoride

Fluoride can cause neurotoxicity in laboratory animals,⁹⁰ but is not shown in the panel as a substance proven to be neurotoxic in man. It exists in drinking water as a natural contaminant, but the concentration is dependent on local geological circumstances. In rural communities in China, high fluoride concentrations in well water might cause skeletal abnormalities. In one such community, 222 children aged 8–13 years showed significantly worse IQs than 290 unexposed controls.⁹⁰ Parallel results were obtained in a smaller study of 118 children of similar age.⁹¹ Another study of 477 schoolchildren from 22 villages suggested that both increased water fluoride concentrations and very low concentrations were associated with IQ deficits, compared with children exposed to normal concentrations (below 1 mg/L).⁹² The reports did not thoroughly consider possible confounders, but do suggest that further in-depth studies be undertaken.

Perchlorate

This chemical is not known as neurotoxic to adults. It is a widespread contaminant of ground water in the USA from the use of ammonium perchlorate as a solid-fuel propellant for rockets and missiles.⁹³ The thyroid is the primary site of perchlorate toxicity, and iodine uptake by the thyroid is blocked. Abnormal brain development, as a consequence of inhibition of maternal thyroid function,^{13,66} is the major potential effect of perchlorate exposure.⁹³

Because the available evidence is uninformative as to neurobehavioral toxicity, drinking water standards for perchlorate are set at levels to protect adults and do not include child-protective safety factors.

Effects of developmental neurotoxicity

The five substances recognised as causes of developmental neurotoxicity show similar patterns in the development of scientific documentation of their risks. This pattern of discovery started in each instance with recognition of adult neurotoxicity, typically in people with occupational exposure and of episodes of acute, high-dose poisoning in children to the next stage was the accumulation of epidemiological evidence of neurobehavioural deficits in children with prenatal exposures at concentrations that are not toxic to adults (figure 1). For lead, methylmercury, and PCBs, widespread subclinical neurotoxicity has been documented internationally, yet the full implications of exposure to arsenic and toluene are unclear. For most substances listed in the panel, only neurotoxicity in adults has been documented.

The combined evidence suggests that neurodevelopmental disorders caused by industrial chemicals has created a silent pandemic in modern society. Although these chemicals might have caused impaired brain development in millions of children worldwide, the profound effects of such a pandemic are not apparent from available health statistics. Additionally, as shown by this Review only a few chemical causes have been recognised so the full effects of our industrial activities could be substantially greater than recognised at present.

As is shown by the evidence for inorganic lead, globally increased exposures have been responsible for erosion of cognitive skills with subclinical, but permanent decreases in IQ. Additionally, this neurotoxic chemical produces lifelong changes in behaviour with shortened attention span, increased impulsivity, heightened aggressiveness, slowed motor coordination, and impaired memory and language skills. The consequences are increased likelihood of school failure, diminished economic productivity, and possibly increased risk of antisocial and criminal behaviour.⁹⁴ The most striking of these effects occur at the extremes of exposure; in highly exposed children, almost none had above average function, whereas the number with obvious deficits increased greatly.⁹⁵ The most severely affected individuals will probably need special education and will also be less likely than their peers to pursue productive career options. A study of adults who were exposed to excess lead as children revealed that they were much less successful in life than those from a less exposed comparison group.⁹⁶

The consequences of a pandemic of developmental neurotoxicity extend beyond descriptive data for incidence and prevalence of clinically diagnosed disorders.^{1,3} Increased risk of Parkinson's disease⁹⁷ or other neurodegenerative diseases⁹⁸ is a further potential consequence

of the pandemic. Thus, early subclinical chemical injury has been postulated to silently kill a fraction of the cells needed to sustain brain function in later life (eg, in the substantia nigra). These latent impairments cause no symptoms in childhood, but could be unmasked during the natural neuronal attrition associated with ageing.^{99,100}

The wide extent of human exposure to pollutants is now becoming apparent after systematic collection of data for the amount of these substances present in the environment and in human tissues.¹⁰¹ Even then, recognition of causal associations could be difficult because exposures vary with time, more than one substance could have an effect, individual vulnerability varies, and there are other factors that can bias epidemiological studies toward the null hypothesis, especially when the outcome might be unrecognised for several years, or even decades.¹⁰²

The population at risk of subclinical neurotoxicity from industrial chemicals is very large. Almost all children born in industrialised countries between 1960 and 1980 were exposed to substantial amounts of lead from petrol that could have reduced the number of children with far above average intelligence (IQ scores above 130 points) by over 50% and might likewise have increased the number with IQ scores below 70.⁹⁵ In the USA alone, the aggregate population of children at risk of exposure to airborne lead at that time was about 100 million. In this period, the resulting economic costs are estimated to have ranged from US\$110 billion to \$319 billion in each year's birth cohort.¹⁰³ Most of these costs were related to the diminished economic productivity that resulted over the exposed children's entire lifetimes from wide-scale reductions in intelligence. Today the costs of lead poisoning are estimated to be \$43 billion in each birth cohort in the USA,⁵ whereas the costs of prenatal methylmercury toxicity are estimated to amount to \$8.7 billion yearly (range, \$2.2–43.8 billion).⁶ Diminished economic productivity remains the main source of these costs. Because of the absence of dose-response associations for other neurotoxic compounds, the total costs are unknown.

The effect of chemical neurotoxicity extends beyond the industrially developed nations. Toxic chemicals, such as highly dangerous pesticides that are banned in industrialised countries, are exported to developing societies, where environmental and occupational standards are often weak or at least poorly enforced.¹⁰⁴ The consequences are largely unreported.

Prevention

A pandemic of neurodevelopmental toxicity caused by industrial chemicals is, in theory, preventable. Testing of new chemicals before allowing them to be marketed is a highly efficient means to prevent toxicity, but has been required only in recent years. Of the thousands of chemicals used in commerce, fewer than half have been subjected to even token laboratory testing for toxicity

testing.²⁴ Nearly 3000 of these substances are produced in quantities of almost 500 000 kg every year, but for nearly half these high-volume chemicals no basic toxicity data are publicly available, and 80% have no information about developmental or paediatric toxicity.²⁴ Although new chemicals must be tested more thoroughly, access to these data can be restricted, because they could be claimed to constitute confidential business information. Absence of information about the neurotoxic potential of most industrial chemicals is therefore the main impediment to prevention of developmental disorders induced by neurotoxic pollutants. Accelerated testing of chemicals already in commerce is therefore essential. In the USA, a legal mandate to require testing was established in the Toxic Substances Control Act, but is largely unenforced.²⁴ In the EU, opportunity exists to require more extensive chemical testing through the REACH programme,²⁵ although the proposed legislation does not emphasise testing for developmental neurotoxicity as a primary objective.

Toxicity testing protocols for chemicals need to be expanded to include examination of neurobehavioural functions. Present test protocols rely mainly on crude indices, such as brain weight and gross morphology.^{105,106} There is a risk that abbreviated protocols used for toxicity screening will overlook neurodevelopmental toxicity, and further testing could erroneously be thought unnecessary. Procedures for functional appraisal are available,¹⁰⁵ and a harmonised protocol for assessment of developmental neurotoxicity was developed under OECD auspices in 1999,¹⁰⁶ although a revision is still under review.

The number of chemicals that can cause neurotoxicity in laboratory studies probably exceeds 1000, which is far more than the estimated 200 that have caused documented human neurotoxicity. However, in the absence of systematic testing,²⁸ the true extent of the neurotoxic potential of industrial chemicals is unknown. The physiology of brain development¹²⁻¹⁴ and experimental evidence^{14,26,27} suggest that developmental neurotoxicity is likely for all of them, except perhaps for some of the compounds that require metabolic transformation to become neurotoxic, in which immature metabolism may provide some degree of protection.^{19,107} The few substances proven to be toxic to human neurodevelopment should therefore be viewed as the tip of a very large iceberg (figure 2).

Large-scale, prospective epidemiological studies, such as birth cohorts from Europe¹⁰⁸ and the National Children's Study proposed in the USA, will be especially informative about early exposures and neurodevelopmental disorders.¹⁰⁹ Data from these investigations, especially when pooled internationally, will hopefully provide dose-response associations that can guide future disease prevention efforts. This research should move beyond repeated assessments of known neurotoxins to examine chemicals, whose toxicity is just beginning to be recognised. The substances listed in the panel, especially

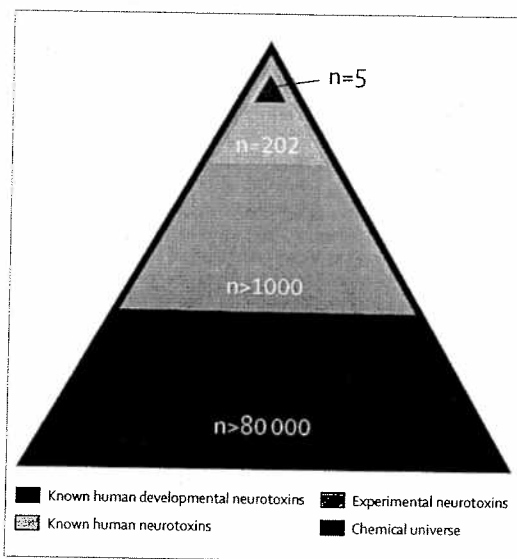


Figure 2: Diagram of the extent of knowledge of neurotoxic chemicals
Of the thousands of known chemicals, only a small fraction have been proven to cause developmental neurotoxicity in humans. Although this evidence does not represent the true potential for industrial chemicals to cause neurodevelopmental disorders, assessments of need for preventive measures nonetheless rely on that information.

those most prevalent in food, drinking water, and the environment, should provide a useful starting point. Nevertheless, these initiatives could take decades to generate the type of detailed documentation required for chemicals regulation.

The Food Quality Protection Act in the USA requires that pesticide standards be set at values that will protect infants against developmental toxicity. If testing data are not available, a child-protective safety factor should be used in standard settings. However, application of this factor has been uneven, and regulatory authorities need to recognise the vulnerability of prenatal brain development.

Prevention of neurodevelopmental disorders of chemical origin will need new approaches to control chemical exposures. The vulnerability of the human nervous system and its special susceptibility during early development suggest that protection of the developing brain should be a paramount goal of public health protection. The high level of proof needed for chemical control legislation has resulted in a slow pace of interventions to prevent exposures to lead and other recognised hazards. Instead, exposure limits for chemicals should be set at values that recognise the unique sensitivity of pregnant women and young children, and they should aim at protecting brain development. This precautionary approach, which is now beginning to be used in the EU, would mean that early indications of a potential for a serious toxic effect, such as developmental neurotoxicity, should lead to strict

regulation, which could later be relaxed, should subsequent documentation show less harm than anticipated.¹⁰ As physicians, we should use prudence when counselling our patients, especially pregnant mothers, about avoidance of exposures to chemicals of unknown and untested neurotoxic potential.

Conflict of interest statement

P Grandjean has testified on behalf of the Natural Resources Defense Council in a court case in regard to mercury pollution from a chemical plant in Maine, USA. PJ Landrigan has testified on behalf of the State of Rhode Island, USA, in a lawsuit against the manufacturers of lead-based paint.

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